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Erratum

Erratum to “On the exciplex formation in the energy transfer from aromatic hydrocarbons to crystal violet”

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The authors regret some interchanges between the set of figures 4&6 and 5&7 (excluding captions) and omission of inset in Fig. 4. The corrected pairing among the figures and captions are given overleaf.

[☆]PII of original article: S1010-6030(97)00263-3

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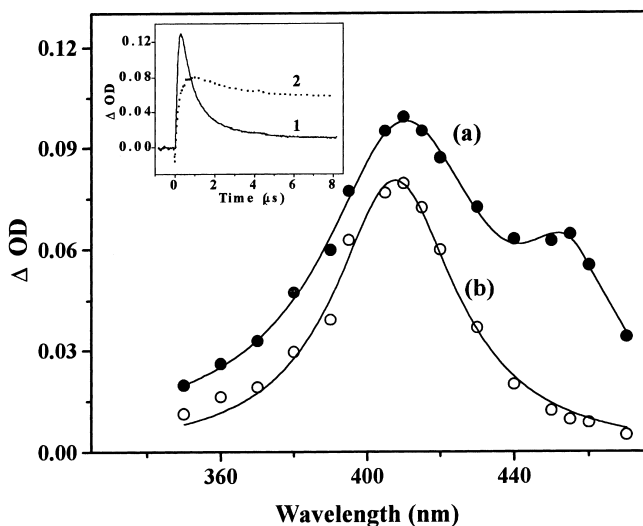


Fig. 4. (a) and (b) are the transient absorption spectra recorded in deoxygenated solutions containing $1.5 \times 10^{-2} \text{ mol dm}^{-3}$ *p*-terphenyl and $3 \times 10^{-4} \text{ mol dm}^{-3}$ CV at 1 μs and 8 μs after the electron pulse respectively. Dose per pulse = 15.8 Gy. Inset: Decay of *p*-terphenyl triplet at 450 nm (1) and build up of $^3\text{CV}^*$ at 400 nm (2).

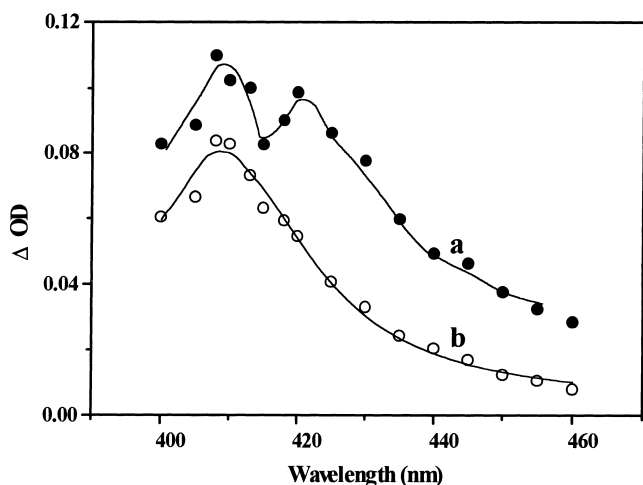


Fig. 6. (a) and (b) are the transient absorption spectra recorded in deoxygenated solutions containing $1.5 \times 10^{-2} \text{ mol dm}^{-3}$ anthracene and $3 \times 10^{-4} \text{ mol dm}^{-3}$ CV at 1 μs and 8 μs after the electron pulse respectively. Dose per pulse = 15.8 Gy.

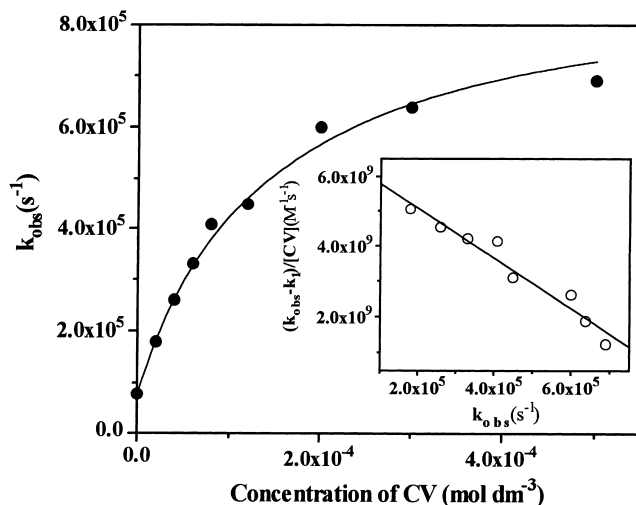


Fig. 5. Decay of *p*-terphenyl triplet (k_{obs}) obtained at 440 nm in deoxygenated solution containing $1.5 \times 10^{-2} \text{ mol dm}^{-3}$ *p*-terphenyl with varying concentrations of CV (●). The solid line represents the calculated curve generated according to equation (14). Inset shows the linear plot obtained as per equation (15).

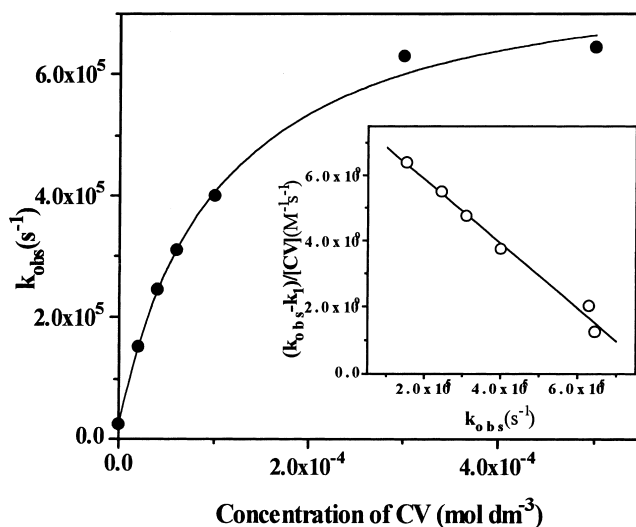


Fig. 7. Decay of anthracene triplet (k_{obs}) obtained at 430 nm in deoxygenated solution containing $1.5 \times 10^{-2} \text{ mol dm}^{-3}$ anthracene with varying concentrations of CV (●). The solid line represents the calculated curve generated according to equation (14). Inset shows the linear plot obtained as per equation (15).